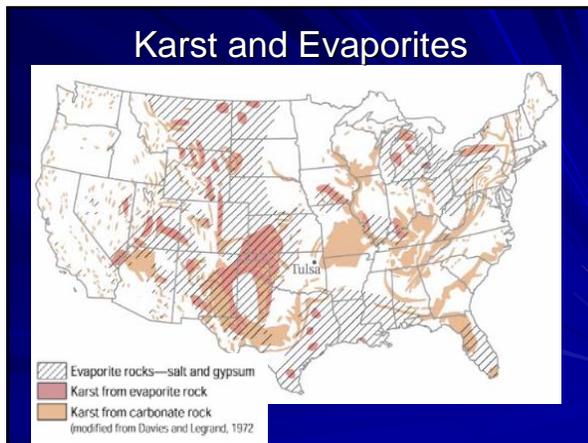


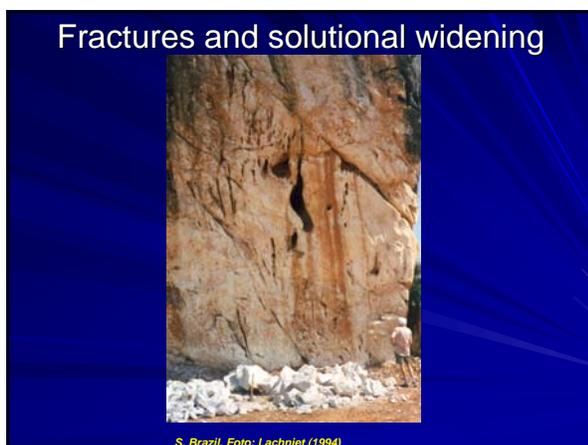
Karst

- Karst is “terrain with distinctive hydrology and landforms arising from a combination of high rock solubility and well-developed secondary porosity”. (Bloom, p. 148)
- Most soluble rock is limestone
- Other rocks also contain minor karst



Lithology and Structure

- Karst dominates in fine-grained, pure (>95% calcite) limestones
- Primary porosity in most limestones is generally low
- Secondary porosity important
 - Joints, bedding planes, fractures, faults
 - Intersection of vertical joint sets promote downward movement of groundwater to form **shafts**



Solution Equations I

- Step 1:
 - $\text{CO}_2 + \text{H}_2\text{O} \rightarrow \text{H}_2\text{CO}_3$ (carbonic acid)
 - Where does CO_2 come from for rainwater?
- Step 2:
 - $\text{H}_2\text{CO}_3 \rightarrow \text{H}^+ + \text{HCO}_3^-$
 - carbonic acid dissociates into hydrogen ion and bicarbonate
- pH of rainwater = ???

CO₂ Sources

- “Aggressiveness” of solution is proportional to [CO₂]
- Atmospheric CO₂
 - Aerated aqueous solution can dissolve 63 mg/L of calcite
- Biologic CO₂
 - [CO₂] in soil zone many orders of magnitude larger than in the atmosphere
 - Soil water solution can dissolve up to 700 mg/L of calcite
 - Most groundwaters contain 200-400 mg/L of calcite

Solution Equations II

- Step 3:
 - $\text{CaCO}_3 \leftrightarrow \text{Ca}^{2+} + \text{CO}_3^{2-}$
 - Calcite dissociation
- Step 4:
 - $\text{H}^+ + \text{CO}_3^{2-} \rightarrow \text{HCO}_3^-$
- Step 5: Putting it all together
 - $\text{CaCO}_3 + \text{H}_2\text{O} + \text{CO}_2 \leftrightarrow \text{Ca}^{2+} + 2\text{HCO}_3^-$

Solution chemistry

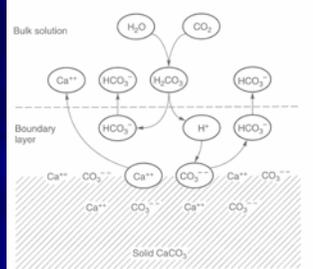
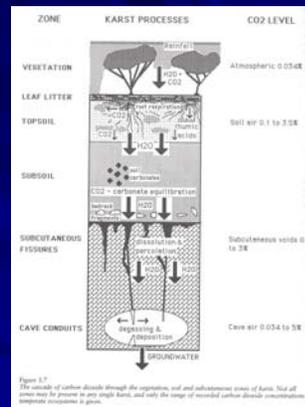


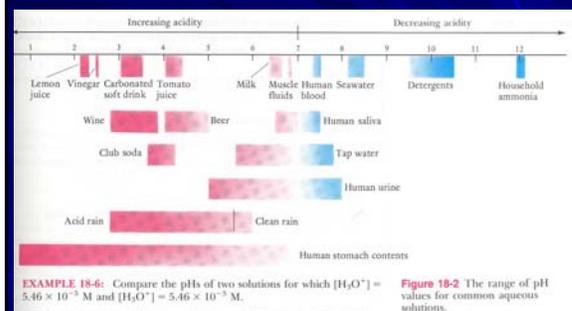
FIGURE 7-13 Steps in the solution of calcite in CO₂-bearing water. (From W. B. White, 1988, Geomorphology and Hydrology of Karst Terrains. Copyright 1988 by Oxford University Press, Inc.)



From Gillespie, D. Caves: Processes, Development, Management

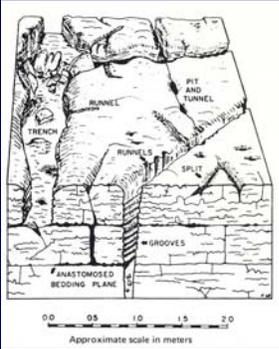
Solution continued

- Cold waters are more soluble to CO₂, thus are potentially more aggressive than warm waters
- But, soil respiration of CO₂ is greatest in hot and humid tropical climates
 - Tropics and mid-latitudes are most prone to karst formation
 - Arctic areas: not enough biologic respiration
 - Deserts: not enough water or biologic respiration



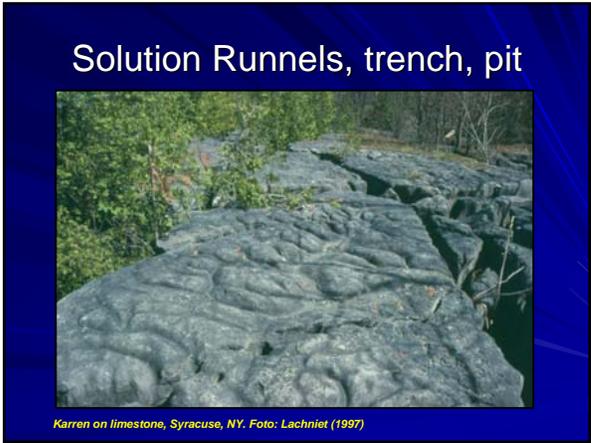
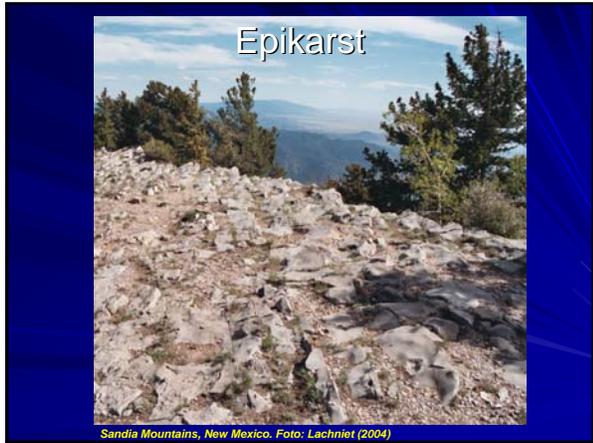
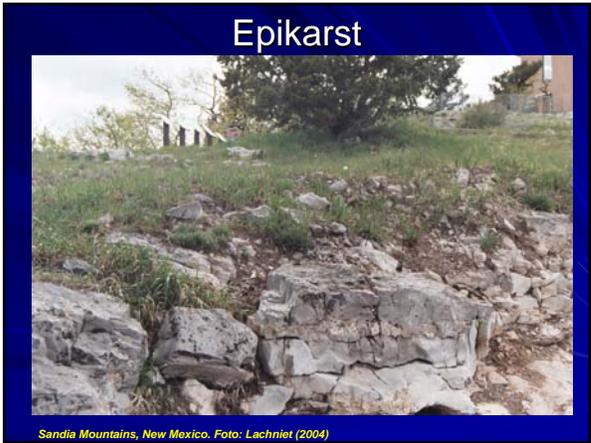
EXAMPLE 18-6: Compare the pHs of two solutions for which [H₃O⁺] = 5.46 × 10⁻⁵ M and [H₃O⁺] = 5.46 × 10⁻⁷ M. Figure 18-2 The range of pH values for common aqueous solutions.

Epikarst landforms



- **Epikarst:**
- **Grikes:** solutionally widened vertical joints
- **Clints:** remnant pavement surface in between grikes
- Rills and Runnels, a.k.a. "**karren**"
- "**Spongework**" dissolution

From Riley, 1991, *Geomorphology*, 3rd Edition, Prentice Hall



Epikarst solution



Sandia Mountains, New Mexico. Foto: Lachniet (2004)

Rillenkarren (or Karren)



Karren, solution features on limestone, Guerrero State, Mexico. Foto: Lachniet (2005)

Karren



Karren, solution features on limestone, Guerrero State, Mexico. Foto: Lachniet (2005)

Major Karst Landforms

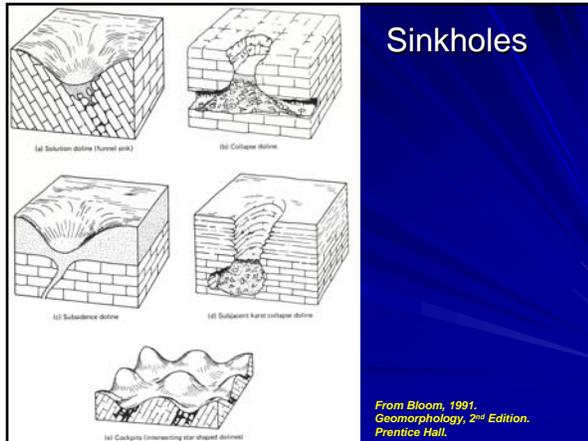
- Fundamental karst landform is the **sinkhole** (aka “doline”)
 - Solutionally enlarged shaft that serves as a conduit from surface to subsurface
 - Note: most “sinkholes” reported by the media **are not true karst sinkholes!**

Solution sinkholes

- Formed by dissolution along joints
- Near vertical
- Funnel-shaped
- Most common on
 - flat terrain and elevated plateaus lacking surface drainage

Collapse Sinkholes

- Near-vertical walls
- Roof collapse over solution cavities
- Rapid collapse
 - Often associated with groundwater lowering
 - Filled with blocks of overlying bedrock
 - Common in Florida



Sinkholes



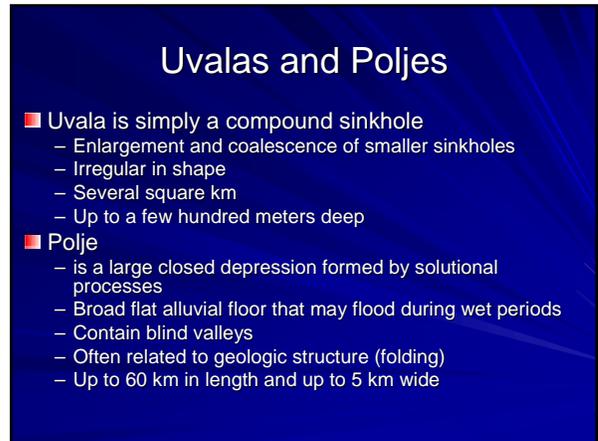
Winter Park, FL sinkhole

- Collapse sinkhole
- Florida is predominantly limestone
- High rainfall and warm temperatures
- Active karst formation



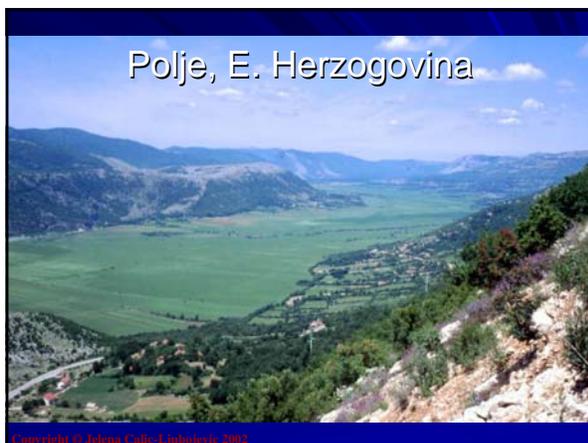
■ Collapse Sinkhole, South Australia

Copyright © Ken Gemen 2003



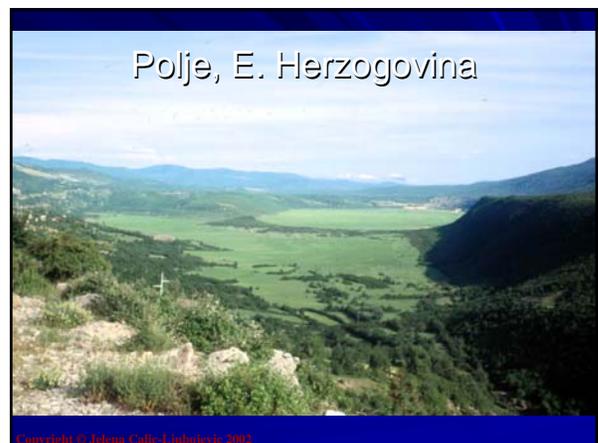
Uvalas and Poljes

- Uvala is simply a compound sinkhole
 - Enlargement and coalescence of smaller sinkholes
 - Irregular in shape
 - Several square km
 - Up to a few hundred meters deep
- Polje
 - is a large closed depression formed by solutional processes
 - Broad flat alluvial floor that may flood during wet periods
 - Contain blind valleys
 - Often related to geologic structure (folding)
 - Up to 60 km in length and up to 5 km wide



Polje, E. Herzegovina

Copyright © Adam Collis-Libbey 2003



Polje, E. Herzegovina

Copyright © Adam Collis-Libbey 2003

Polje in Sierra Cuchumatanes, Guatemala



Polje solution on limestone, Guatemala. Foto: Lachniet (2006)

Swallow Hole



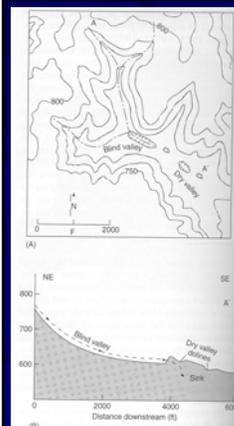
Copyright © Don Harmon 2003

Swallow Hole



Copyright © Randy Schaefer 2007

Blind Valley



- A sinking stream flows into groundwater system
- Upstream is called a blind valley
- Downstream is called a dry valley

Epigenic and Hypogenic Caves

- Epigenic:
 - Movement of water from overlying or adjacent recharge surfaces
 - 90% of caves
 - Atmospheric and biologic CO_2 source for aggressive waters
- Hypogenic:
 - Deep-seated water sources
 - 10% of caves
 - Hydrosulfuric acid (H_2S) and igneous carbonic (H_2CO_3) acid
 - Carlsbad Caves, New Mexico

Cave Formation

- Form in 10^4 to 10^5 years (typo p. 205)
- A cave is a solutional opening large enough for a person to enter
- Consist of
 - Passages – longer than they are high
 - Rooms – higher than wide

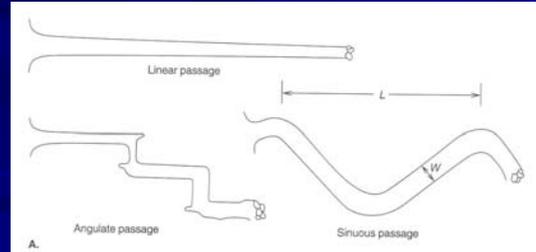
Caves are big enough for a person to fit inside



Death Valley National Park, CA. Photo by Stephen Hlowjski, 2004

Single-conduit Passages

- May have dendritic drainage pattern
- One passage develops more rapidly than others, effectively capturing more water

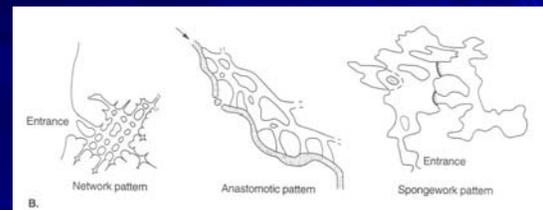


Sinuious cave stream



Foto: Keith Christenson

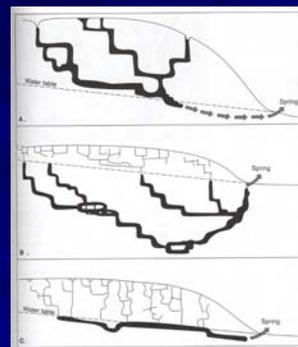
Maze Passages



Three principle theories of Cave origins

- 1) Form above water table by **vadose water** (unsaturated zone water)
- 2) Form beneath water table by circulation of **phreatic water** (=saturated zone water)
- 3) Form at water table = **water table caves**

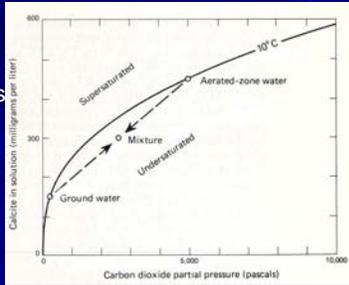
Theories of cave formation



- 1) Vadose Zone
- 2) Phreatic Zone
- 3) Water Table Cave

Mixing Effect

- Mixing two saturated solutions in equilibrium with two $[CO_2]$ produces an undersaturated solution!
- Thus, mixing of different groundwaters will result in additional solution



Water Table Cave, Panama



Foto: Keith Christenson

Domingo's Gullet



Foto: Keith Christenson

Water Table Cave

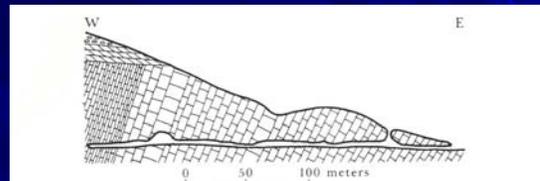
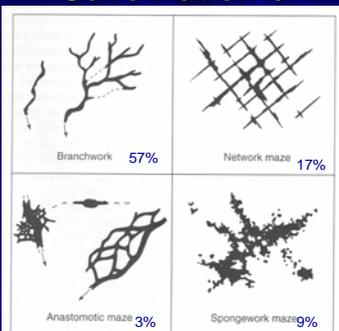


Figure 7-15. Simplified cross section of Lehman Caves, Nevada. Horizontal passages intersect tilted strata. (Diagram courtesy of G. W. Moore.)

From Bloom, 1991. *Geomorphology*, 2nd Edition. Prentice Hall.

Cave Patterns

Cave equivalent of dendritic drainage



Commonly form along a low angle fracture

Angular grids of intersecting fissures
Diffuse gw flow

May originate from solutional opening of primary porosity; Hypogenic

Figure 12.25 Single-passage caves 14%
Branchwork and maze patterns of cave development.
(Pulver 1991.)

Residual Karst

- Cockpit karst**
 - leaves conical limestone hills separated by star-shaped sinkholes
 - Best developed in humid tropics
 - Sinkhole distribution probably controlled by geologic structure
- Tower Karst**
 - Steep to vertical limestone hills
 - Swampy plains

Cockpit Karst in Guatemala



Copyright © Matthew Lachniet, 2007

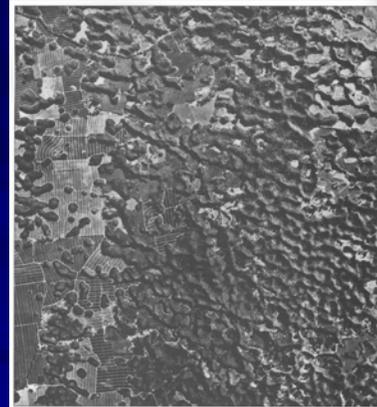
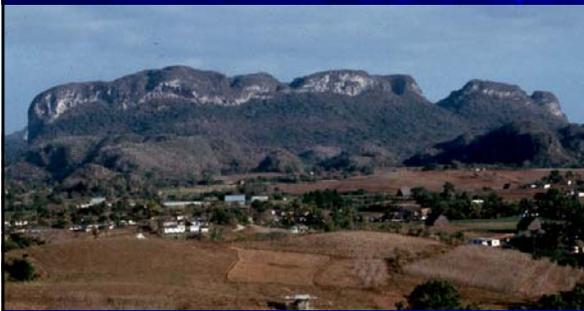


Figure 12.20
Aerial view of cockpit karst.

Tower Karst, Cuba



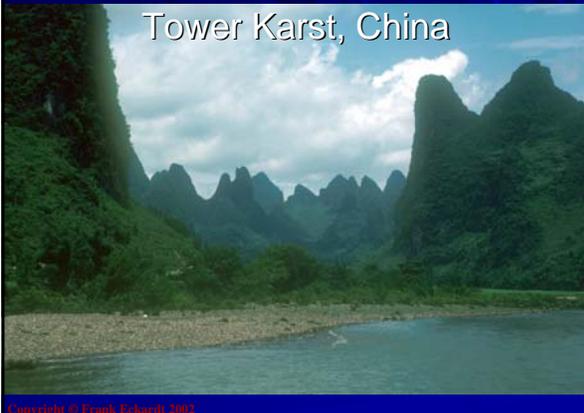
Copyright © Don C. Dugan, 2002

Tower Karst, Costa Rica



Guanacaste Peninsula, NW Costa Rica. Foto: Lachniet (1999)

Tower Karst, China



Copyright © Frank Eckert, 2002

Speleothems

- Cave waters dissolve calcite under high $[CO_2]$ derived from the soil zone
- $CaCO_3 + H_2O + CO_2 \leftrightarrow Ca^{2+} + 2HCO_3^-$
 - $[CO_2]$ in cave passages = atmospheric
 - Saturated drip waters enter cave
 - CO_2 is released
 - Above equation proceeds in reverse
 - calcite is precipitated into amazing cave calcite formations called **speleothems**

